Cytotoxic Naphthoquinones from Roots of Lippia microphylla

Hélcio S. Santos^a, Sônia M. O. Costa^a, Otília D. L. Pessoa^a, Manoel O. Moraes^b, Claudia Pessoa^b, Sergio Fortier^b, Edilberto R. Silveira^a, and Telma L. G. Lemos^{a,*}

- ^a Departamento de Química Orgânica e Inorgânica, Universidade Federal do Ceará, CP 12200, 60451-970 Fortaleza, Ceará, Brazil E-mail: tlemos@dqoi.ufc.br
- b Departamento de Fisiologia e Farmacologia, Universidade Federal do Ceará, CP 3157, 60430-270 Fortaleza, Ceará, Brazil
- * Author for correspondence and reprint requests
- Z. Naturforsch. 58 c, 517-520 (2003); received December 9, 2002/February 20, 2003

A new prenylated naphthoquinone dimer named microphyllaquinone (1), a mixture of 6-methoxy- and 7-methoxy-naphtho[2,3-b]-furan-4,9-quinones (2a/2b) and tecomaquinone I (3), were isolated from roots of *Lippia microphylla*. The structures were elucidated by spectroscopic methods, including detailed 1D and 2D (COSY, NOESY, HMQC, HMBC) NMR data. Unpublished ¹³C NMR data of 2a and 2b are reported. The *in vitro* cytotoxic activity of the isolated compounds was tested against five types of tumor cells.

Key words: Lippia microphylla, Naphthoquinones, Cytotoxicity Activity

Introduction

The genus Lippia (Verbenaceae) constituted of herbs, shrubs and small trees, includes approximately 200 species, distributed mainly in South and Central America and Tropical Africa territories. Most of them are traditionally utilized in folk medicine for the treatment of several diseases (Lemos et al., 1990; Valentin et al., 1995; Forestiere et al. 1996; Pascual et al., 2001). As a continuation of our program on the evaluation of plants used in Brazilian traditional medicine, we have investigated L. sidoides (Lemos et al., 1999; Costa et al., 2001) and volatile components of the aerial parts of L. microphylla (Lemos et al., 1992). In the present paper we report the isolation and characterization of a new prenylated naphthoguinone dimer (1), a mixture of isomers 6-methoxy- and 7-methoxy-naphtho[2,3-b]-furan-4,9-quinones (2a and 2b, respectively) and tecomaguinone I (3), from the roots of L. microphylla, popularly known as "alecrim-de-tabuleiro". Since 2a/2b had been previously isolated from *Lantana camara*, only the ¹H NMR data were provided (Abeygunawardena et al., 1991). Here the complete ¹H and ¹³C NMR spectral data are reported for the first time.

Results and Discussion

After chromatography over silica gel investigation of the ethanol extract from the roots of *L*.

$$\begin{matrix} \mathbf{R_1} & \mathbf{0} & \mathbf{O} \\ \mathbf{0} & \mathbf{0} & \mathbf{O} \\ \mathbf{R} & \mathbf{0} & \mathbf{0} \\ \mathbf{0} & \mathbf{0} & \mathbf{0} \\ \mathbf{0} & \mathbf{0} & \mathbf{0} \\ \mathbf{0} & \mathbf{0} & \mathbf{0} \\ \end{matrix}$$

2a R = OMe, R₁ = H 2b R = H, R₁ = OMe

Fig. 1. Structures of microphyllaquinone (1); 6-methoxy/7-methoxy-naphtho[2,3-*b*]-furan-4,9-quinones, (2a/2b); tecomaquinone I (3).

microphylla, led to the isolation of four naphthoquinones. Three of them were the known furanonaphthoquinones 2a/2b and tecomaquinone I (3). Compound 1, was isolated as red crystals, m.p. 197-200° C. Its IR spectrum showed absorption bands for hydroxyl (3403 cm⁻¹) and conjugated carbonyl groups (1665 and 1640 cm⁻¹). The molecular ion peak with m/z 440 daltons, in conjunction with the ¹H and ¹³C NMR spectra (Table I) showing 27 signals, allowed the determination of its molecular formula as C₂₇H₂₀O₆. The DEPT 135° spectrum revealed three methyl groups, one as methoxy (δ 52.17), one methine, nine monohydrogenated sp² carbons, and by comparison with the ¹³C NMR spectrum fourteen non-hydrogenated sp² carbons, including three carbonyls (δ 184.8, 182.5 and 171.2) and two oxygenated (δ 154.8 and 147.8). ¹H and ¹³C NMR data comparison of **1** and tecomaquinone I (3) (Lemos et al., 1999), revealed a striking similarity, except for the missing cyclic isoprene moiety for 1 and in addition, the presence of a methyl ester for 1, Fig. 1. Even though a more deshielded absorption for C-3 (conjugated with the methyl ester) was expected, initially the complete ¹³C data assignment was a difficult task because the absence of any hydrogen long-range correlation with C-3 or C-2 (this one is ortho to the hydroxy, but is conjugated to the quinone C-1') did not allow an unambiguous assignment of those carbons. Fortunately, the HMBC data obtained in dry DMSO-d₆ revealed long-range correlations of the hydroxy hydrogen (δ 10.88) with C-1 (δ 154.8), C-9 (δ 127.7) and C-2 (δ 104.4). In addition, the distinction between C-2'(δ133.4) and C-3' $(\delta 137.4)$, both conjugated to the quinone carbonyls, was done by the correlations of the carbinolic like, and doubly allylic H-11'(δ 6.51) with C-4 (δ 147.8), C-2' (δ 133.4), C-3' (δ 137.4) and C-1' $(\delta 182.5)$, as well as by the correlations of the vinyl hydrogen H-12' (δ 5.6) with C-14'(δ 26.4) and C-15'(δ 19.3). Finally, the relative stereochemistry suggested for 1, similarly to tecomaquinone I (3), was confirmed by the NOESY analysis. Thus the structure of 1, a new prenylated naphthoquinone

Table I. ¹³C NMR (125 MHz) and ¹H NMR (500 MHz) data for 1.

С	δ_{C}	$_{\mathrm{HMQC}}^{\mathrm{HMQC}}$ $_{\mathrm{H}}^{\mathrm{J}}$ (Hz)	${}^{2}\!J_{\rm CH}$ HMBC	$^3 \! J_{ m CH}$
1	154.8			H-8
2	104.4			
3	109.8			
4	147.8			H-11', H-5
5	123.5	8.27 d $(J = 9,5)$		H-7
6	129.8	7.69 m		H-8
2 3 4 5 6 7 8	129.0	7.69 m		H-5
	124.7	8.44 d (J = 9.5)		H-6
9	127.7			H-5, H-7
10	128.6			H-6, H-8
11	171.2			MeO-11'
1'	182.5			H-11', H-8'
2'	133.4			H-11'
3'	137.4			H-11'
4'	184.8	0.44 1 (1 0.7)		H-5'
5'	126.8	8.11 d (J = 8.7)		H-7'
6'	134.0	7.78 m		H-8'
7'	134.1	7.78 m		H-5'
8' 9'	126.5	8.16 d (J = 7.5)		H-6'
-	132.6			
10'	133.5	(51 1 (1 02)	II 10/	
11'	68.3	6.51 d $(J = 9.3)$	H-12'	211 14/ 211 15/
12' 13'	117.9 142.8	5.60 d (J = 9.3)	H-11'	3H-14′, 3H-15′
14'	26.4	1.68 s	3H-14′, 3H-15′	H-11'
14 15'	19.3	2.10 s		H-12', 3H-15' H-12', 3H-14'
MeO-11	52.2	2.10 s 3.72 s		11-12 , 311-14
HO-1	34.4	11.28 s		
110 1		11.20 6		

dimer named microphyllaquinone was established.

Compounds 2a/2b, were obtained as orange crystals, m.p. 162-165° C. Its IR spectrum showed strong absorption bands consistent with the presence of a conjugated carbonyl (1675 cm⁻¹) and aromatic ring (1591, 1580 and $1480 \,\mathrm{cm}^{-1}$). The molecular formula C₁₃H₈O₄ was deduced from its ¹H and ¹³C NMR spectra. All proton and carbon signals appeared in pairs (except to the OMe group) with light chemical shift differences. This clearly suggested that 2 should be a mixture of isomers. Integration of the ¹H NMR signals allowed to establish a 2:1 relative ratio of **2a/2b**. Full assignments of ¹H and ¹³C NMR spectra of the major isomer were assisted by COSY, HMQC and HMBC experiments and are consistent with the structure of 6-methoxynaphtho[2,3-b]-furan-4,9-quinone (2a). Therefore, the minor compound must be 7-methoxy-naphtho[2,3-b]-furan-4,9-quinone (2b). Direct comparison of the ¹H NMR data of these compounds with the literature values showed to be identical (Abeygunawardena et al., 1991).

Compounds 1 and 2a/2b were tested against five cultured tumor cell lines *in vitro*. These substances display cytotoxicity values ranging from 0.8 to 3.0 µg/ml when tested against HL-60, CEM, HCT8 and MCF-7 human cancer cell lines obtained from the American Type Culture Collection (ATCC) and B16 (murine melanome) from Universidade Federal do Rio de Janeiro. Compound 1 was slightly less active than the furanonaphthoquinones 2a/2b which showed higher cytotoxicity against resistant cell lines B16 and HCT-8. The IC₅₀ values of each component for the proliferation of five tumor cells are summarized in Table II. The cytotoxic activity of tecomaquinone I (3) was previously reported (Costa *et al.*, 2001).

Experimental

General procedures

Melting points were measured on a Mettler Toledo FP90 apparatus and were uncorrected. The optical rotations were measured on a Perkin-Elmer 341 digital polarimeter. IR spectra were recorded on a Perkin-Elmer 1000 FT-IR spectrometer. Mass spectral data were acquired on a VG Auto Spec spectrometer. The NMR spectra were recorded in CDCl₃ and DMSO-d₆ on a Bruker Avance DRX-500 (500 MHz for ¹H and 125 MHz for ¹³C) spectrometer. Proton and carbon chemical shifts were referenced relative to the corresponding solvents signals, δ_H 7.24 and δ_C 77.0 (CHCl₃) and $\delta_{\rm H}$ 2.49 and $\delta_{\rm C}$ 39.5 (DMSO- d_6). Si gel 60 (Merck, 70-230 mesh) was used for column chromatography. Precoated Si gel plates (Merck, kieselgel 60 F₂₅₄, 0.20 mm) were used for analytical TLC. Chromatographic fractions were visualized by TLC by spraying with vanillin-perchloric acid-EtOH followed by heating.

Plant material

The roots of *L. microphylla* were collected in Varzea Alegre, Ceará State – Brazil, in April, 1999. The identity of plant was verified by Dr. Edson P. Nunes, and a voucher specimen (No. 2287) was deposited in the Herbarium Prisco Bezerra of the Departamento de Biologia, Universidade Federal do Ceará.

Extraction and isolation

The dried, ground biomass (940 g) from the roots of *L. microphylla* was extracted at room temperature with EtOH. After solvent evaporation under reduced pressure, a brown residue

Table II. IC_{50} values in $\mu g/ml$ from cytotoxicity assay^a of naphthoquinones from L. microphylla.

	IC ₅₀ (mean ± SE)	
Cell Lines	(1)	(2a/2b)
B16 (murine melanoma) CEM (human limphocyte leukemia) HL60 (human promyelocyte leukemia) HCT8 (human colon adenocarcinoma) MCF7 (human breast adenocarcinoma)	3.13 ± 0.12 2.57 ± 0.33 2.92 ± 0.13 2.33 ± 0.29 2.44 ± 0.13	0.77 ± 0.05 1.61 ± 0.25 1.62 ± 0.09 0.69 ± 0.06 1.30 ± 0.05

 $^{^{\}rm a}$ Taxol (0.14 $\mu g/ml)$ was used as the positive control (100% death of all cell lines).

(30 g) was obtained. This material was fractioned over a Si gel column with hexane/CHCl₃ (1:1 v/v), CHCl₃, EtOAc and MeOH. The CHCl₃ fraction (8.2 g) was chromatographed over Si gel and eluted with 0-100% EtOAc-hexane mixtures. The residue obtained from the 5% EtOAc-hexane fraction was subjected to repeated column chromatography to yield 1 (100 mg), 3 (15 mg, m.p. 188-189° C, according previous reported data), and, 2a/2b (8 mg) as an isomer mixture.

Cytotoxicity bioassays

For CEM and HL-60 (leukemia), B16 (melanoma), HCT8 (colon) and MCF-7 (breast) cancer cell lines, a microassay for cytotoxicity was performed using the microtitre MTT (tetrazonium [3-(4'-5'-dimethylthiazol-2'-yl)-2,5-diphenyl-tetrazoliumbromide]) dye assay under conditions of continuous drug exposure as previously described (Mosmann, 1983), by the following procedure. Adherent cancer cell lines at the concentration of 0.3×10^6 cells/ml and suspended cells at 0.5×10^6 cells/ml were seeded in 96 well microplates. The adherent cells were incubated during twenty four hours to allow cell attachment. The compounds were added to the cell cultures at concentrations of 0.39–25 µg/ml, and the cells incubated for three days. The MMT solution was added three hours before the end of the incubation time. Cell survival was evaluated with a multiwell scanning spectrophotometer at 540 nm. All compounds were tested in three replicate wells in each plate, and experiments were repeated at least two times.

Microphyllaquinone (1): $C_{27}H_{20}O_6$, red crystals, m.p. 197–200° C; $[\alpha]_D^{25}$ –130.2 (c 0.004, CHCl₃); IR (KBr): v_{max} = 3403, 2913, 1665, 1640, 1595, 1439, 1372, 1335, 1241, 1034, 850, 769,704 cm⁻¹; ¹H (500 MHz) and ¹³C (125 MHz) NMR data, see Table I; EIMS (70 eV) m/z 440 ([M]⁺, 47), 425 (19), 408 (13), 393 (100), 381 (6).

Methoxynaphtho[2,3-b] furan-4,9-quinones (2): $C_{13}H_8O_4$, orange crystals, m.p. $162-165^{\circ}$ C; IR (KBr): $v_{\text{max}} = 2923$, 1675, 1591, 1480, 1439, 1363, 1268, 1233, 1098, 969, 883, 774 cm⁻¹. ¹H NMR for **2a** (500 MHz, CDCl₃): $\delta = 3.99$ (s, OMe), 7.00 (d, J = 1.5, H-3), 7.22 (dd, J = 8.7, 2.5, H-7), 7.69 (d, J = 2.5, H-5), 7.77 (d, J = 1.5, H-2), 8.15 (d, J =8.7, H-8); ¹³C NMR for **2a** (125 MHz, CDCl₃): 56.4 (s, OMe), 109.2 (C-3), 111.5 (C-5), 119.9 (C-7), 126.9 (C-8a), 129.9 (C-8), 131.2 (C-3a), 135.1 (C-4a), 149.0 (C-2), 153.1 (C-9a), 164.7 (C-6), 173.9 (C-4), 180.2 (C-9). ¹H NMR for **2b** (500 MHz, CDCl₃): $\delta = 3.99$ (s, OMe), 6.99 (d, J = 1.5, H-3), 7.20 (dd, J = 8.7, 2.5, H-6), 7.66 (d, J = 2.5, H-6) H-8), 7.75 (d, J = 1.5, H-2), 8.18 (d, J = 8.7, H-5); ¹³C NMR for **2b** (125 MHz, CDCl₃): 56.4 (s, OMe), 108.9 (C-3), 111.8 (C-8), 119.9 (C-6), 126.1 (C-4a), 129.8 (C-5), 130.5 (C-3a), 135.9 (C-8a), 148.5 (C-2), 153.5 (C-9a), 164.7 (C-7), 173.4 (C-4), 180.9 (C-9).

Acknowledgements

This research was financially supported by a grants from the Brazilian National Agencies (CNPq/CAPES and FUNCAP).

- Abeygunawardena C., Kumar V., Marshall D. S., Thomson R. H., and Wickramaratne D. B. M. (1991), Furanonaphthoquinones from two *Lantana* species. Phytochemistry **30**, 941–45.
- Costa S. M. O., Lemos T. L. G., Pessoa O. D. L., Pessoa C., Montenegro R. C., and Braz-Filho R. (2001), Chemical constituents from *Lippia sidoides* and cytotoxic activity. J. Nat. Prod. 64, 792–795.
- Forestiere A. M., Monfort M. T., Ragusa S., and Trovato A. (1996), Antiinflamatory, analgesic and antipyretic activity in rodents of plants extracts used in Africa medicine. Phytother. Res. **10**, 100–106.
- Lemos T. L. G., Matos F. J. A., Alencar J. W., and Craveiro A. A. (1990), Antimicrobial activity of essential oils of Brazilian plants. Phytother. Res. 4, 82–84.
- Lemos T. L. G., Monte F. J. Q., Matos F. J. A., Alencar J. W., and Craveiro A. A. (1992), Chemical composition and antimicrobial activity of essential oil from Brazilian plants. Fitoterapia 63, 266–268.

- Lemos T. L. G., Costa S. M. O., Pessoa O. D. L., and Braz-Filho R. (1999), Total assignments of ¹H and ¹³C NMR spectra of tectol and tecomaquinone I. Magn. Reson. Chem. **37**, 908–911.
- Mosmann T. (1983), Rapid colorimetric assay for cellular growth and survival: application to proliferation and cytotoxic assays. J. Immunol. Methods **65**, 55–63.
- Pascual M. E., Slowing K., Carretero E., Sánchez Mata D., and Villar A. (2001), *Lippia*: traditional uses, chemistry and pharmacology: a review. J. Ethnopharmacol. 76, 201–214.
- Valentin A., Pelissier Y., Benoit F., Marion C., Kone D., Mallie M., Bastide J. M., and Bessiere J. M. (1995), Composition and antimalarial activity in vitro of volatile components of *Lippia multiflora*. Phytochemistry 40, 1439–1442.